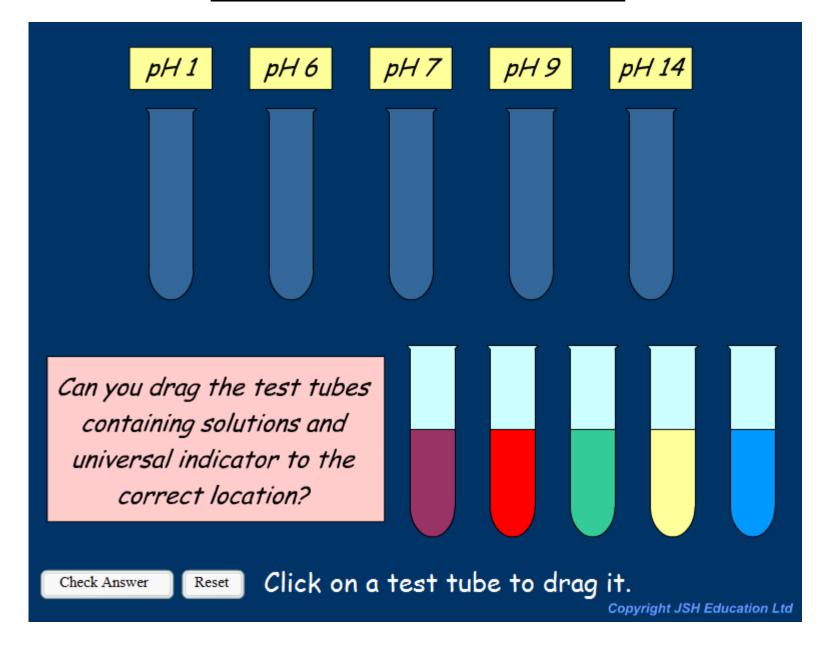


IT: Universal Indicator

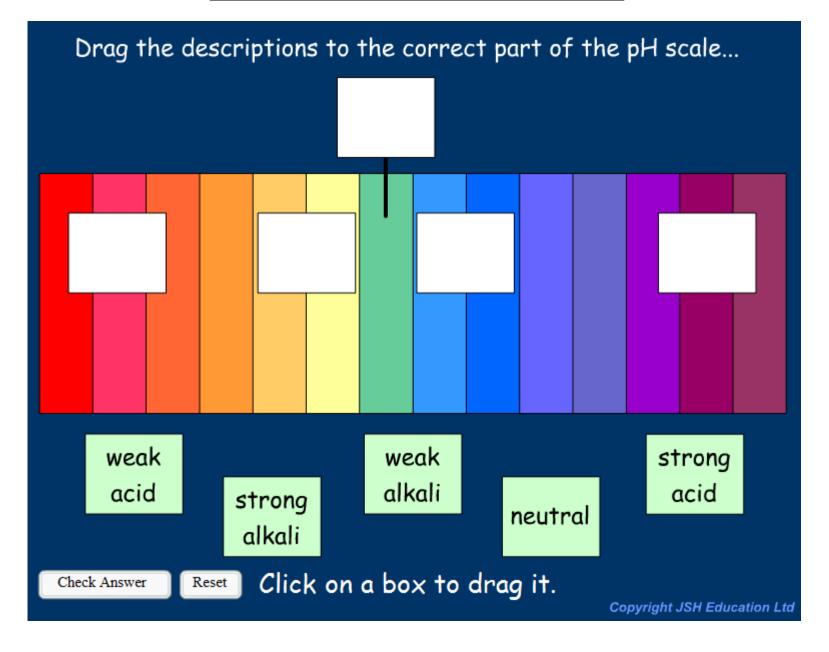






IT: Universal Indicator





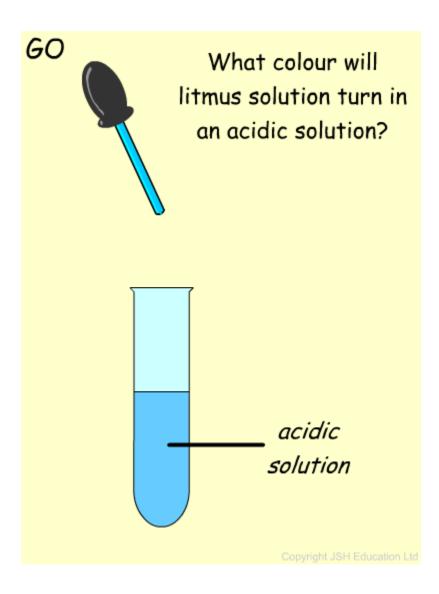








Investigating Litmus











Exercise: Apparatus



Draw a diagram for the apparatus shown in the photograph.

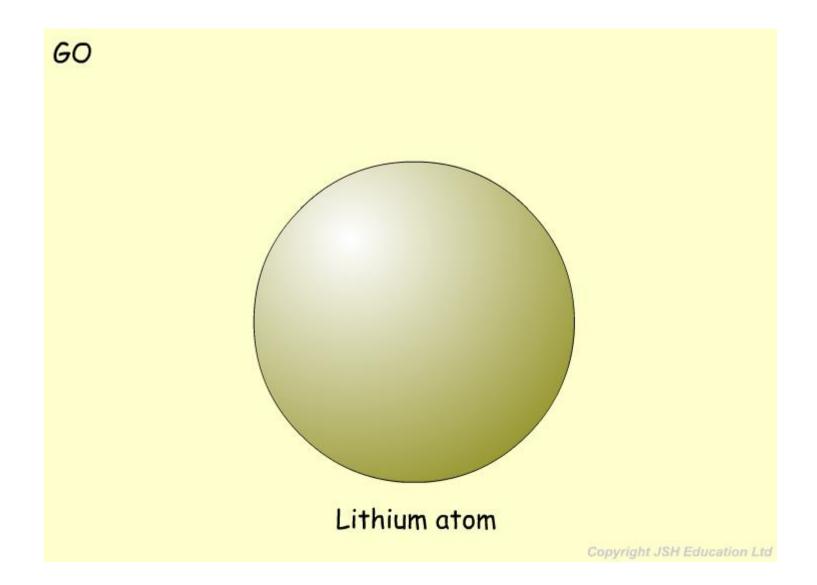








A Model Of The Lithium Atom











Exercise: Particles

- Which three particles make up an atom?
 Neutron, proton and electron
- Which two particles are found in the nucleus?
 Proton and neutron
- 3. Which particle has a positive charge?
 The proton
- **4.** Which particle has a negative charge? The electron
- 5. Which particle orbits the atom?
 The electron









Exercise: Which Elements?

- For the compounds below, write down the names of the elements present.
 - NH_3 a) Nitrogen and hydrogen
 - b) CH₄ Carbon and hydrogen
 - H₂SO₄ Hydrogen, sulfur and oxygen
 - d) CaCl₂ Calcium and chlorine
 - e) CO Carbon and oxygen
 - f) NaCl Sodium and chlorine
 - SiO₂ Silicon and oxygen g)
 - Fe_2O_3 h) Iron and oxygen



IT: Chemical Symbols



Drag the chemical symbol to the correct element.					
Chlorine	Copper		Ag Ca		
Argon	Iron		K Fe		
Potassium	Silver		Cl Ar		
Calcium	Gold		Cu Au		
Check Answer Reset Click on a symbol to drag it. Copyright JSH Education Ltd					









<u>CFCs</u>

CFCs in the atmosphere are known to deplete the ozone layer.

They are a group of **man-made chemicals** that were first produced in the first half of the twentieth century.

CFCs usage includes as coolants in refrigeration systems and air conditioners, as solvents to clean electronic components and in the manufacture of plastic foam.

Countries are now trying to **limit the production of CFCs** because of their adverse effect upon the ozone layer.

Even though worldwide production of CFCs has decreased, because the CFCs remain active in the atmosphere for many years it is expected that the ozone layer will only recover by the middle of the twenty first century.









UK Production Of CFCs



Comment on the information shown.

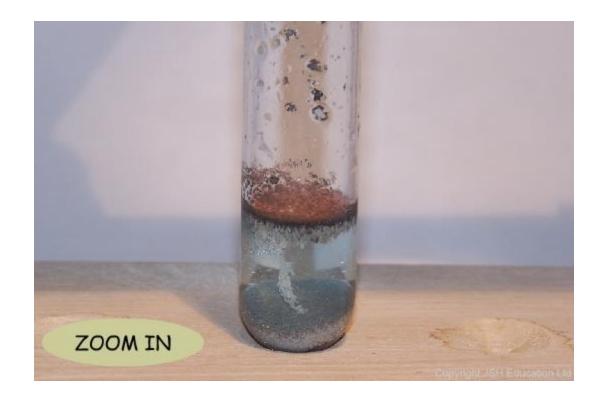








Copper & Silver

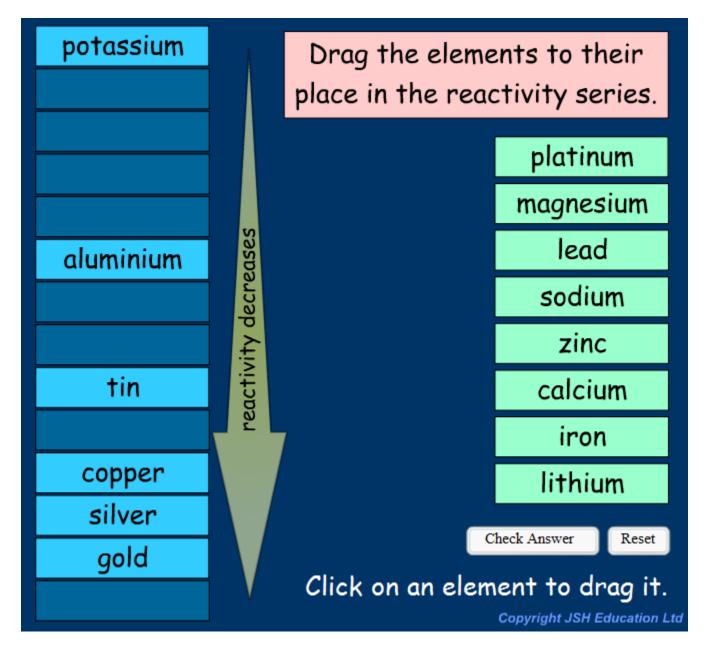


Above you can see some copper powder that has just been added to silver nitrate solution. When copper is added to a solution of silver nitrate, copper will displace the silver from the solution.



IT: The Reactivity Series

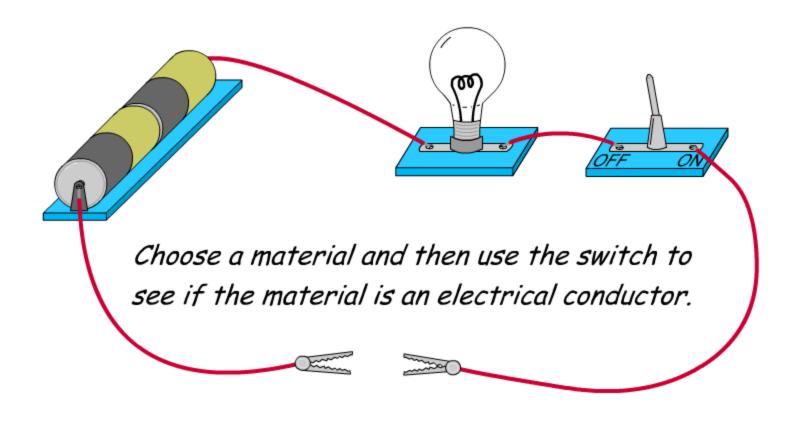








Testing Materials



copper	wood	steel	glass
graphite	plastic	iron	aluminium









Properties Of Metals



Metals share many common properties. Can you name any?





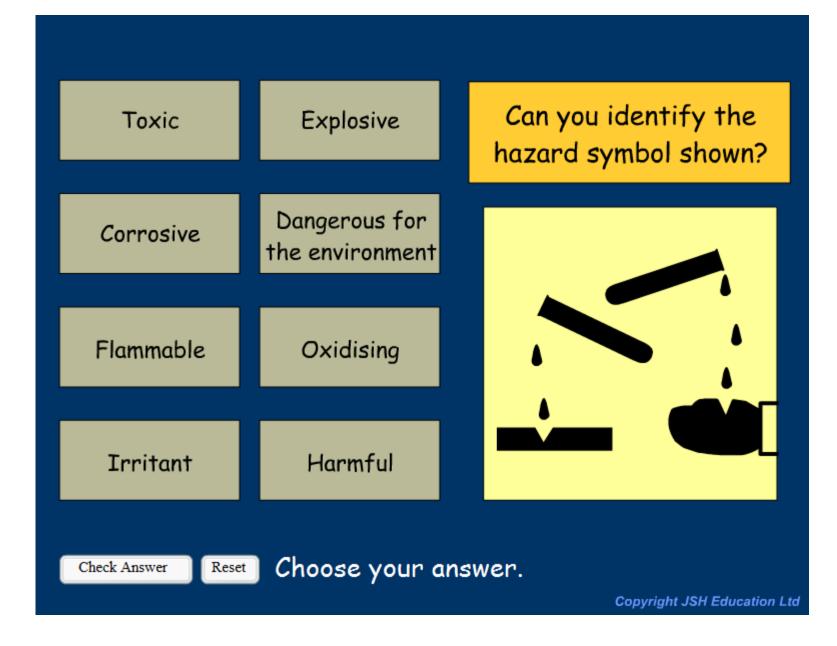
Metamorphic Viewer





IT: Hazard Symbols

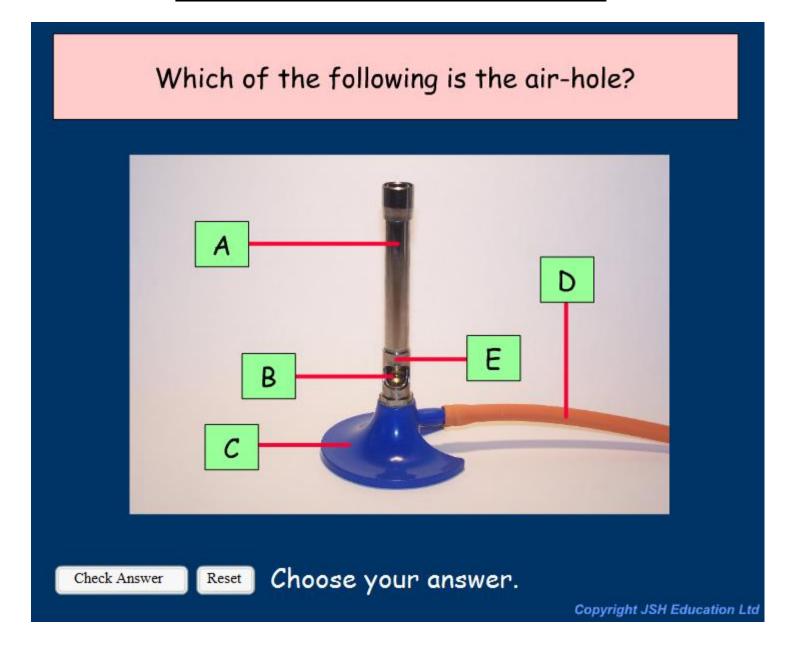






IT: The Bunsen Burner













<u>Settling</u>

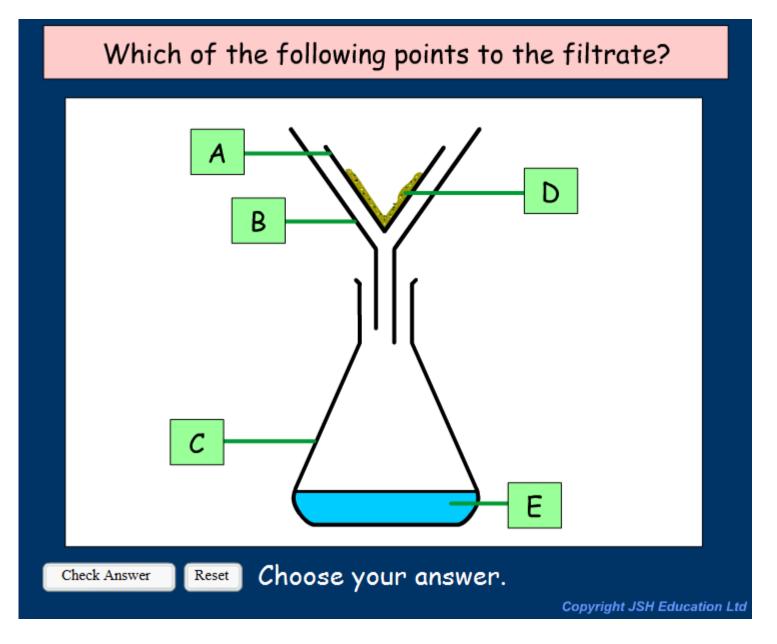


Gravity causes the sand particles to settle to the base of the beaker.











IT: States Of Matter



